Tetrahedron 67 (2011) 904-909

Contents lists available at ScienceDirect

Tetrahedron

journal homepage: www.elsevier.com/locate/tet

First total synthesis of tripdiolide

Bing Zhou, Huanyu Tang, Huijin Feng, Yuanchao Li*

Shanghai Institute of Materia Medica, Chinese Academy of Sciences, 555 Road Zu Chong Zhi, Zhangjiang Hi-Tech Park, Shanghai 201203, PR China

ARTICLE INFO

ABSTRACT

Article history: Received 28 October 2010 Received in revised form 1 December 2010 Accepted 7 December 2010 Available online 10 December 2010

Keywords: Tripdiolide Total synthesis Abietic acid Tripterygium wilfordii Hook. f. α,β-Unsaturated lactone ring

1. Introduction

Tripterygium wilfordii Hook. f. (TWHF), commonly known as Lei Gong Teng (Thunder God Vine), has been used in Traditional Chinese Medicine to treat autoimmune and inflammatory diseases, such as rheumatoid arthritis for centuries.^{1–3} Tripdiolide 1 and Triptolide 2 (Fig. 1), the two major diterpene bioactive components responsible for the clinical properties of TWHF, were first isolated from TWHF extracts and characterized as a diterpenoid triepoxide lactone containing an 18 (4 \rightarrow 3) abeo-abietane skeleton in 1972.⁴ Tripdiolide 1 has also been reported to be effective in the treatment of leukaemia^{4–6} and to have anti-cancer, anti-inflammatory, immunosuppressive and anti-fertility activities.⁷⁻¹¹ In addition, tripdiolide **1** had a wider therapeutic index¹² than most of the other diterpenoids in TWHF, e.g., triptolide 2, whose 14-succinyl sodium salt (PG490-88),¹³ a water-soluble prodrug converted to triptolide in serum, entered into phase I clinical trials in Europe for the treatment of solid tumors in 2003. All these properties mentioned above suggest that tripdiolide **1** should be a promising drug, however, tripdiolide unfortunately remains scarcely accessible from the natural source and the scarcity of this substance (0.0008% of dry root weight)¹⁴ hampered further biological studies. Although synthesis of Triptolide has been reported,¹⁵ there is no publication on the synthesis of Tripdiolide. Herein we described the first total synthesis of tripdiolide **1** by a route utilizing the readily available

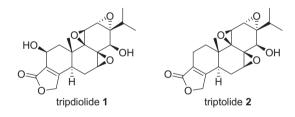


Fig. 1. Tripdiolide and triptolide.

abietic acid **3** as the practical starting material. The results are reported in this paper.

2. Results and discussion

The first total synthesis of tripdiolide **1** from the readily available abietic acid **3** in 22 steps is described

and this synthesis furnishes a 5.8% overall yield of 1 with an average yield of 87%.

Structurally, tripdiolide **1** has α , β -unsaturated lactone ring, three epoxide adducts and C2- β -hydroxyl group and our own interest in **1** grew out of a desire to find an efficient route for its total synthesis, so we chose the very accessible abietic acid **3** as the starting material after analyzing the structure of **1** and **3**. Our retrosynthetic route for compound **1** is depicted in Scheme 1. The crucial steps include the introduction of C2- β -hydroxyl and α , β -unsaturated lactone ring and triepoxide construction from the 2 β -hydroxyl triptophenolide methyl ether **4**.

Synthesis of 2 β -hydroxyl triptophenolide methyl ether **4** started with known abietic acid **3**, which was converted to C4-(18)-alkenes **5** by the known procedure reported by us (Scheme 2).¹⁶ Allylic oxidation of **5** with SeO₂/TBHP provided C-3 α alcohol **6**, which was oxidized readily under Swern conditions¹⁷ to the enone **7** in 83% yield for the two steps. When compound **7** was treated with



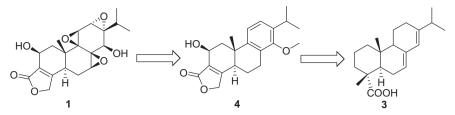


© 2010 Elsevier Ltd. All rights reserved.

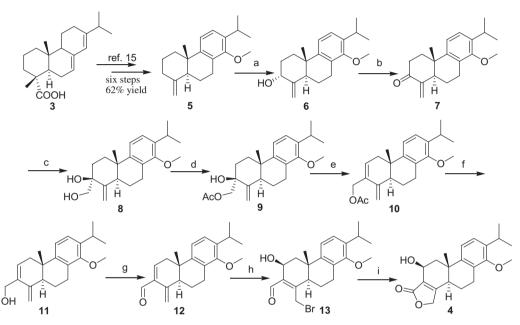
Tetrahedror

^{*} Corresponding author. Tel.: +86 21 50806600x3502; fax: +86 21 50807288; e-mail address: ycli@mail.shcnc.ac.cn (Y. Li).

^{0040-4020/\$ –} see front matter @ 2010 Elsevier Ltd. All rights reserved. doi:10.1016/j.tet.2010.12.014



Scheme 1. Retrosynthetic analysis of Tripdiolide 1 from abietic acid 3.



Scheme 2. Synthesis of 2β -hydroxyl triptophenolide methyl ether 4 from abietic acid 3. Reagents and conditions: (a) SeO₂, t-BuO₂H, CH₂Cl₂, rt, overnight, 88%; (b) DMSO, (COCl₂), Et₃N, CH₂Cl₂, -78 °C, 2 h, 94%; (c) (isopropoxy-dimethylsilyl) methyl chloride, Mg, THF, -30 °C, 55 min; then KF, KHCO₃, 30% H₂O₂, THF, MeOH, rt, overnight, 80% over 2 steps; (d) Ac₂O, Py, rt, 3 h, 98%; (e) SOCl₂, Py, rt, 8 h, 81%; (f) K₂CO₃, MeOH, H₂O, rt, 2 h, 98%; (g) Dess–Martin periodinane, NaHCO₃, CH₂Cl₂, rt, 3 h, 97%; (h) NBS, acetone, H₂O, rt, 4h, 73%; (i) NaClO₂, 2-Methyl-2-butene, NaH₂PO₄, *t*-BuOH, THF, H₂O, rt, 3 h, 85%.

isopropoxy-dimethylsilylmethylmagnesium chloride¹⁸ in THF at -30 °C, nucleophilic addition proceeded smoothly providing β-hydroxysilane intermediate, which without purification, was subjected to oxidative cleavage of the corresponding Si-C bond by potassium fluoride and 30% hydrogen peroxide to give 1,2-diol 8 as a single stereoisomer in 80% yield. Protection of primary hydroxyl group with acetic anhydride provided compound 9, which was treated with SOCl₂ in pyridine, affording **10** in 79% yield for the two steps. The acetyl group can be removed simply from 10 by stirring with K₂CO₃ in MeOH and H₂O, providing compound **11** in 98% yield. Subsequent oxidation of 11 proceeded smoothly with Dess-Martin periodinane to afford diene aldehyde 12, which without purification, upon oxidation of diene with NBS¹⁹ in acetone and H_2O gave 1,4-product **13** in 71% yield for the two steps, along with a small trace of 2α -hydroxyl isomer ($2\beta/2\alpha=15:1$). The results of reaction of diene aldehyde 12 with NBS in different solvents at different temperature are summarized in Table 1. We found that treatment of **12** with NBS in acetone and water (5:1) at room temperature afforded 13 in a higher yield (73%). When aldehyde 13 was oxidized with sodium chlorite,²⁰ lactonization of the resulting bromo acid occurred spontaneously to give the desired 2^β-hydroxyl triptophenolide methyl ether 4 in 85% yield.

With the key intermediate **4** in hand, we switched to the construction of triepoxide. Treatment of **4** with ammonium ceric nitrate²¹ in H₂O/CH₃CN provided the C-7 α alcohol **14** in 81% yield (Scheme 3). However, we were disappointed to find that exposure of compound **14** to a variety of oxidizing agents²² only afforded a small trace of desired ketone **15** and direct oxidation of **4** with

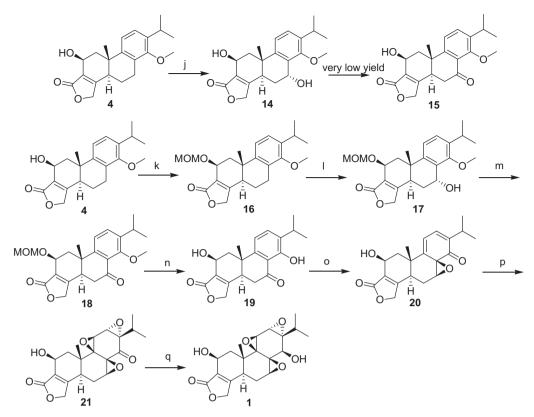
Table 1

The reaction of diene aldehyde 12 with NBS in different solvents at different temperature

Substrate	Solvent	Temperature	Yield ^a
12	DMSO/H ₂ O=1:1	rt	37%
12	DMSO/H ₂ O=5:1	rt	42%
12	Acetone/H ₂ O=1:1	rt	59%
12	Acetone/H ₂ O=5:1	rt	73%
12	THF/H ₂ O=1:1	rt	47%
12	THF/H ₂ O=5:1	rt	56%
12	Acetone/H ₂ O=5:1	0 °C	57%
12	Acetone/H ₂ O=5:1	40 °C	50%

^a Isolated yield of compound **13**.

some oxidizing agents, such as CrO₃/HOAc or sodium dichromate, also failed to give ketone **15**, which proved unstable to a variety of conditions, including silica gel chromatography. Considering that the instability of **15** may be related to the presence of 2β -hydroxyl group, exposure of the key intermediate **4** to MOMCl gave compound **16** in 92% yield. Treatment of **16** with ammonium ceric nitrate in H₂O/CH₃CN provided the alcohol **17** in 89% yield, which upon oxidation of hydroxyl with pyridinium dichromate delivered ketone **18** in 97% yield. As predicted, **18** was a stable intermediate and readily purified by silica gel chromatography. In addition, the MOM and methyl protecting group can be removed simultaneously under the BBr₃ deprotection condition, affording phenol **19** in 97% yield. With the important intermediate **19** in hand, the compound **19** was converted into tripdiolide **1** by a sequence similar to that developed previously for construction of the *C*-ring functionality.¹⁵f



Scheme 3. Synthesis of tripdiolide 1 from 2β-hydroxyl triptophenolide methyl ether 4. Reagents and conditions: (j) (NH₄)₂Ce(NO₃)₆, CH₃CN/H₂O=1/1, rt, 3 h, 81%; (k) MOMCl, N,Ndiisopropylethylamine, CH₂Cl₂, rt, overnight, 92%; (l) (NH₄)₂Ce(NO₃)₆, CH₃CN/H₂O=1/1, rt, 2 h, 89%; (m) PDC, CH₂Cl₂, rt, overnight, 97%; (n) BBr₃, CH₂Cl₂, -78 °C to rt, 97%; (o) NaBH₄, CH₃OH, 0 °C, 1 h; then NaIO₄, CH₃OH/H₂O=3/1, 0 °C, 1 h, 94%; (p) CF₃COCH₃, Oxone, NaHCO₃, CH₃CN/aqueous Na₂(EDTA)=1:1, 25 °C, 4 h; then H₂O₂, NaOH, MeOH, 25 °C, 2 h, 69%; (q) Eu(fod)₃, NaBH₄, CH₃OH, -78 °C, 1 h, 47%.

Treatment of **19** with NaBH₄ in MeOH gave C-7 β alcohol, which without purification, upon oxidation with sodium periodate²³ provided C7,8- β -epoxy dienone **20** in 94% yield for the two steps. When methyl(trifluoromethyl)dioxirane generated in situ was used,²⁴ the second C9,11- β -epoxide was introduced as a single diastereomer. Further epoxidation with basic H₂O₂ provided 2 β -hydroxyl triptonide **21** having triepoxides as the sole epoxidation product in 69% yield for the two steps. Finally, reduction of triptonide **21** with NaBH₄ in MeOH in the presence of Eu(fod)₃^{15f} furnished tripdiolide **1** (47%) together with its C14- α -hydroxyl epimer (45%). The synthesized tripdiolide **1** and compound **21** showed the identical NMR spectra and optical rotation to the data reported in the literature.²⁵

3. Conclusion

In summary, we have established an efficient route for the synthesis of tripdiolide **1** from readily accessible abietic acid **3**. Notably, this synthesis furnishes a 5.8% overall yield of **1** with an average yield of 87% and features an NBS-mediated oxidation of **12** followed by treatment with sodium chlorite to spontaneously install the α , β -unsaturated lactone ring and the C2 stereogenic hydroxyl group with excellent stereocontrol. More importantly, it will allow rapid access to various tripdiolide analogues designed to probe the cellular processes that tripdiolide interferes with.

4. Experimental section

4.1. General

Mass spectra and high-resolution mass spectra were measured on a Finnigan MAT-95 mass spectrometer. Melting points were determined on a Buchi 510 melting point apparatus and are uncorrected. IR spectra were recorded on a Nicollet Magna FT-IR-750 spectrometer using KBr pellets. Optical rotations were recorded on a Jasco-Dip-181 polarimeter. ¹H and ¹³C NMR spectra were determined on Bruker AM-300, Bruker AM-400 instruments using tetramethylsilane as internal reference. Data are presented as follows: chemical shift, multiplicity (s=singlet, br s=broad singlet, d=doublet, br d=broad doublet, t=triplet, m=multiplet), *J*=coupling constant in hertz (Hz). The signals of the ¹³C NMR were assigned utilizing DEPT experiments and on the basis of literature data. Silica gel 60H (200–300 mesh) manufactured by Qingdao Haiyang Chemical Group Co. (China) was used for general chromatography.

4.1.1. (2R,4aS,10aR)-7-isopropyl-8-methoxy-4a-methyl-1-methylene-1,2,3,4,4a,9,10,10a-octahydrophenanthren-2-ol (6). To a solution of compound 5 (13 g, 0.045 mol) in CH₂Cl₂ (500 mL) were added t-BuOOH (70 wt % in water, 19.1 mL, 0.135 mol) and SeO₂ (2.5 g, 0.022 mol). The mixture was stirred at room temperature overnight. NaHSO₃ (5 g) was added and the mixture was washed with brine. The organic phase was dried over Na₂SO₄ and concentrated to give a crude product, which was chromatographed on silica gel (5% EtOAc in *n*-hexane) to give pure 6 (11.9 g, 88%) as a colourless solid, ¹H NMR (CDCl₃, 300 MHz) δ 7.07 (s, 2H), 5.08 (s, 1H), 4.76 (t, *J*=1.8 Hz, 1H), 4.36 (s, 1H), 3.73 (s, 3H), 3.30 (sept, *I*=3.0 Hz, 1H), 3.14–3.04 (m, 1H), 2.80–2.67 (m, 2H), 2.08–1.60 (m, 7H), 1.23 (d, *J*=3.0 Hz, 3H), 1.21 (d, *J*=2.4 Hz, 3H), 0.98 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 154.8, 151.6, 145.7, 138.2, 128.4, 123.6, 121.3, 109.8, 72.7, 60.4, 41.1, 39.0, 32.4, 30.1, 26.0, 24.0, 23.9, 23.8, 21.8, 20.3; LRMS (EI, 70 eV) m/z (%) 300 (M⁺, 10), 267 (100); HRMS (EI) calcd for C₂₀H₂₈O₂ (M⁺): 300.2090, found 300.2086.

4.1.2. (4aS,10aR)-7-isopropyl-8-methoxy-4a-methyl-1-methylene-3,4,4a,9,10,10a-hexahydrophenanthren-2(1H)-one (**7**). To a solution

907

of DMSO (7.1 mL) in CH₂Cl₂ (200 mL) were added (COCl)₂ (4.9 mL) at -78 °C under nitrogen and the mixture was stirred at -78 °C for 30 min. To the resultant solution was added 6 (15 g, 0.05 mol) in CH₂Cl₂ (100 mL). The solution was stirred at -78 °C for 40 min and Et₃N (36 mL) was added dropwise. The solution was stirred for 30 min, warmed to room temperature, and water (20 mL) was added dropwise. The mixture was washed with water and brine. The organic phase was dried over Na₂SO₄ and concentrated to give a crude product, which was chromatographed on silica gel (2% EtOAc in *n*-hexane) to give pure **7** (14 g, 94%) as a colourless solid, ¹H NMR (CDCl₃, 300 MHz) δ 7.11 (s, 2H), 5.98 (s, 1H), 5.23 (s, 1H), 3.74 (s, 3H), 3.31 (sept, J=3.9 Hz, 1H), 3.16 (dd, J=18.0, 4.2 Hz, 1H), 2.78–2.45 (m, 5H), 2.14–1.90 (m, 2H), 1.74–1.60 (m, 1H), 1.24 (d, J=3.9 Hz, 3H), 1.21 (d, J=3.9 Hz, 3H), 1.13 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 202.1, 154.8, 148.3, 144.0, 138.7, 128.4, 124.1, 121.7, 118.4, 60.5, 46.1, 37.2, 36.7, 36.1, 26.1, 24.0, 23.8, 23.8, 22.2, 20.3; LRMS (EI, 70 eV) *m/z* (%) 298 (M⁺, 78), 241 (100); HRMS (EI) calcd for C₂₀H₂₆O₂ (M⁺): 298.1932, found 298.1940.

4.1.3. (2S,4aS,10aR)-2-(hydroxymethyl)-7-isopropyl-8-methoxy-4amethyl-1-methylene-1,2,3,4,4a,9,10,10a-octahydrophenanthren-2-ol (8). A solution of 7 (0.298 g, 1.0 mmol) in THF (6.0 mL) was added to the Grignard reagent [prepared from chloromethyldimethylisopropoxysilane (0.627 mL, 3.5 mmol), 1,2-dibromoethane (two drops), and Mg (0.096 g, 4.0 mmol) in THF (4.0 mL) according to the Tamao's procedure¹⁸ under Ar atmosphere. After stirring at –30 °C for 55 min, the mixture was quenched with an aqueous NH₄Cl solution (10%) and extracted with EtOAc. The organic layer was washed with brine and dried over Na₂SO₄, and concentrated to give a single adduct as colourless oil. To a stirred mixture of colourless crude adduct, MeOH (5.0 mL), THF (5.0 mL), KHCO₃ (0.150 g, 1.5 mmol), and KF (0.282 g, 3.0 mmol) was added H₂O₂ (30%, 0.5 mL, 5.0 mmol) at room temperature. The mixture was stirred at room temperature until starting material disappeared. An aqueous Na₂S₂O₃ solution (50%) was added slowly to the mixture and stirred until a negative starch/iodide test was observed. The mixture was extracted with EtOAc. The organic layer was washed with brine, dried over Na₂SO₄, and concentrated. The residue was purified via column chromatography (10% EtOAc in *n*-hexane) to provide compound $\mathbf{8}$ (0.264 g, 80%) as a white solid, mp 158–160 °C; $[\alpha]_D^{25}$ +273 (*c* 0.15, CH₂Cl₂); IR (KBr) 3388, 2960, 2935, 2867, 1060, 1031 cm⁻¹, ¹H NMR (CDCl₃, 300 MHz) δ 7.06 (m, 2H), 5.38 (s, 1H), 4.99 (s, 1H), 3.76-3.72 (m, 4H), 3.55 (d, J=11.1 Hz, 1H), 3.30 (sept, J=3.0 Hz, 1H), 3.15-3.06 (m, 1H), 2.78-2.64 (m, 1H), 2.46 (br s, 1H), 2.26–2.19 (m, 2H), 2.08–1.56 (m, 6H), 1.23 (d, J=2.7 Hz, 3H), 1.21 (d, J=3.0 Hz, 3H), 1.02 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 154.9, 151.1, 145.2, 138.5, 128.3, 123.9, 121.4, 108.6, 75.6, 66.8, 60.4, 44.5, 39.0, 35.6, 33.0, 26.0, 24.2, 23.8, 23.8, 22.3, 21.2; LRMS(EI, 70 eV)m/z(%) 330(M⁺, 15), 299 (100); HRMS (EI) calcd for C₂₁H₃₀O₃ (M⁺): 330.2195, found 330.2187.

4.1.4. ((2S,4aS,10aR)-2-hydroxy-7-isopropyl-8-methoxy-4a-methyl-1-methylene-1,2,3,4,4a,9,10,10a-octahydrophenanthren-2-yl)methyl acetate (9). To a solution of compound 8 (4.2 g, 0.013 mol) in pyridine (20 mL) was added Ac₂O (12 mL, 0.13 mol). The mixture was stirred at room temperature for 3 h and CH₂Cl₂ was added. The organic layer was washed with 5% HCl and brine, dried over Na₂SO₄, and concentrated to give a crude product, which was chromatographed on silica gel (3% EtOAc in *n*-hexane) to give pure **9** (4.7 g, 98%) as a colourless oil, ¹H NMR (CDCl₃, 300 MHz) δ 7.05 (m, 2H), 5.36 (s, 1H), 4.93 (s, 1H), 4.34 (d, J=11.7 Hz, 1H), 4.15 (d, J=11.1 Hz, 1H), 3.73 (s, 3H), 3.30 (sept, J=3.6 Hz, 1H), 3.10 (dd, J=17.4, 5.7 Hz, 1H), 2.80-2.65 (m, 2H), 2.32-1.62 (m, 10H), 1.23 (d, *J*=3.6 Hz, 3H), 1.21 (d, *J*=3.6 Hz, 3H), 1.01 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 171.2, 154.9, 150.4, 145.2, 138.5, 128.3, 123.8, 121.4, 108.5, 74.8, 68.4, 60.4, 44.5, 38.9, 35.6, 32.9, 26.0, 24.3, 23.9, 23.8, 22.2, 21.2, 20.7; LRMS (EI, 70 eV) m/z (%) 372 (M⁺, 27), 299 (100); HRMS (EI) calcd for $C_{23}H_{32}O_4$ (M⁺): 372.2300, found 372.2302.

4.1.5. ((4aS,10aS)-7-isopropyl-8-methoxy-4a-methyl-1-methylene-1,4,4a,9,10,10a-hexahydrophenanthren-2-yl)methyl acetate (10). To a solution of compound 9 (70 mg, 0.188 mmol) in pyridine (0.5 mL) was added SOCl₂ (0.034 mL 0.47 mmol). The mixture was stirred at room temperature for 8 h and CH₂Cl₂ was added. The organic laver was washed with 5% NaHCO3 and brine, dried over Na2SO4, and concentrated to give a crude product, which was chromatographed on silica gel (0.5% EtOAc in *n*-hexane) to give pure **10** (53.9 mg, 81%) as a colourless solid, ¹H NMR (CDCl₃, 300 MHz) δ 7.11–7.02 (m, 2H), 5.95 (d, J=4.5 Hz, 1H), 5.12 (s, 1H), 5.03 (s, 1H), 4.88 (d, J=12.9 Hz, 1H), 4.69 (d, *J*=12.6 Hz, 1H), 3.73 (s, 3H), 3.31 (sept, *J*=6.0 Hz, 1H), 3.14 (dd, *J*=17.4, 5.1 Hz, 1H), 2.72–2.56 (m, 2H), 2.44–2.36 (m, 2H), 2.20–2.10 (m, 1H), 2.09 (s, 3H), 1.67–1.51 (m, 1H), 1.23 (d, J=6.3 Hz, 3H), 1.21 (d, *J*=6.0 Hz, 3H), 0.99 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 171.0, 154.5, 144.8, 144.1, 138.3, 132.2, 129.2, 128.6, 124.0, 121.9, 107.7, 65.1, 60.5, 43.9, 41.0, 36.8, 26.1, 23.9, 23.9, 23.8, 23.1, 21.0, 20.1; LRMS (EI, 70 eV) *m*/*z* (%) 354 (M⁺, 56), 279 (100); HRMS (EI) calcd for C₂₃H₃₀O₃ (M⁺): 354.2195, found 354.2197.

4.1.6. ((4aS,10aS)-7-isopropyl-8-methoxy-4a-methyl-1-methylene-1,4,4a,9,10,10a-hexahydrophenanthren-2-yl)methanol (11). To a solution of compound 10 (354 mg, 1.0 mmol) in MeOH (20 mL) and water (5 mL) was added K₂CO₃ (552 mg, 4.0 mmol). The mixture was stirred at room temperature for 3 h and the solvent was evaporated, then AcOEt was added and the mixture was washed with brine. The organic layer was dried over Na₂SO₄ and concentrated to give a crude product, which was chromatographed on silica gel (10% EtOAc in *n*-hexane) to give pure **11** (305 mg, 98%) as a colourless solid, mp 122–124 °C; $[\alpha]_D^{25}$ +246 (*c* 0.105, CH₂Cl₂); IR (KBr) 3421, 2962, 2935, 2869, 2836, 1648, 1608, 1033 cm⁻¹, ¹H NMR (CDCl₃, 400 MHz) δ 7.10–7.03 (m, 2H), 5.90 (d, J=5.2 Hz, 1H), 5.20 (s, 1H), 5.03 (s, 1H), 4.46-4.26 (m, 2H), 3.73 (s, 3H), 3.30 (sept, J=6.8 Hz, 1H), 3.16–3.09 (m, 1H), 2.70–2.58 (m, 2H), 2.42–2.35 (m, 2H), 2.18–2.11 (m, 1H), 1.63–1.50 (m, 2H), 1.22 (d, J=6.8 Hz, 3H), 1.21 (d, J=6.8 Hz, 3H), 0.98 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 154.5, 145.0, 144.4, 138.2, 136.5, 128.6, 126.3, 124.0, 121.9, 107.2, 63.9, 60.5, 44.1, 40.9, 36.9, 26.0, 24.0, 23.9, 23.8, 23.2, 20.1; LRMS (EI, 70 eV) m/ *z* (%) 312 (M⁺, 100); HRMS (EI) calcd for C₂₁H₂₈O₂(M⁺): 312.2090, found 312.2093.

4.1.7. (3S,4aS,10aR)-1-(bromomethyl)-3-hydroxy-7-isopropyl-8-methoxy-4a-methyl-3,4,4a,9,10,10a-hexahydrophenanthrene-2-carbaldehyde (13). To a solution of compound 11 (330 mg, 1.06 mmol) in CH₂Cl₂ (20 mL) were added NaHCO₃ (266 mg, 3.18 mmol) and Dess-Martin periodinane (672 mg, 1.59 mmol). The mixture was stirred at room temperature for 3 h and water (20 mL) was added. The mixture was washed with water and brine. The organic phase was dried over Na₂SO₄ and concentrated to give a crude product **12**, which without purification, was dissolved in acetone (20 mL) and water (4 mL) and was added NBS (198 mg, 1.11 mmol). The mixture was stirred at room temperature for 4 h and the solvent was evaporated, then AcOEt was added and the mixture was washed with brine. The organic layer was dried over Na₂SO₄ and concentrated to give a crude product, which was chromatographed on silica gel (10% EtOAc in *n*-hexane) to give pure **13** (305 mg, 71% over two steps) as a colourless solid, ¹H NMR (CDCl₃, 400 MHz) δ 10.3 (s, 1H), 7.16–7.10 (m, 2H), 4.74 (d, *J*=6.4 Hz, 1H), 4.70 (d, *J*=10.8 Hz, 1H), 4.37 (d, J=10.8 Hz, 1H), 3.73 (s, 3H), 3.29 (sept, J=7.2 Hz, 1H), 3.26-3.13 (m, 2H), 2.89-2.79 (m, 1H), 2.67-2.59 (m, 2H), 2.42-2.36 (m, 1H), 1.92-1.80 (m, 2H), 1.41 (s, 3H), 1.21 (d, J=7.2 Hz, 3H), 1.19 (d, J=7.2 Hz, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 191.7, 155.3, 154.9, 144.2, 138.8, 137.1, 127.7, 124.2, 121.0, 63.1, 60.5, 45.0, 41.6, 35.1, 26.1, 24.2, 24.2, 24.2, 23.9, 23.8, 19.3; LRMS (EI, 70 eV) m/z (%) 406 (M⁺, 30), 309 (100); HRMS (EI) calcd for $C_{21}H_{27}BrO_3~(M^+)$: 406.1144, found 406.1153.

4.1.8. 2β -Hydroxyl triptophenolide methyl ether (**4**). To a solution of compound **13** (25 mg, 0.06 mmol) in THF (3 mL), *t*-BuOH (3 mL) and H₂O (1 mL) were added 2-methyl-2-butene (0.25 mL), NaH₂₋ PO₄.2H₂O (47 mg, 0.3 mmol) and NaClO₂ (27 mg, 0.3 mmol). The mixture was stirred at room temperature for 3 h and the solvent was evaporated, then AcOEt was added and the mixture was washed with brine. The organic layer was dried over Na₂SO₄ and concentrated to give a crude product, which was chromatographed on silica gel (15% EtOAc in *n*-hexane) to give pure 4 (17.4 mg, 85%) as a colourless solid, mp 192–194 °C; $[\alpha]_{D}^{25}$ +30 (c 0.06, CH₂Cl₂); IR (KBr) 3477, 2962, 2931, 1749, 1675, 1033 cm⁻¹, ¹H NMR (CDCl₃, 300 MHz) δ 7.14 (m, 2H), 4.88 (m, 2H), 4.75 (d, J=7.5 Hz, 1H), 3.73 (s, 3H), 3.29 (sept, *J*=3.9 Hz, 1H), 3.14–2.85 (m, 2H), 2.75–2.60 (m, 2H), 2.04–1.90 (m, 3H), 1.24 (d, J=3.9 Hz, 3H), 1.21 (d, J=3.9 Hz, 3H), 1.18 (s, 3H); 13 C NMR (CDCl₃, 100 MHz) δ 173.7, 165.2, 155.3, 143.8, 139.1, 127.5, 126.6, 124.3, 120.7, 70.6, 60.5, 60.3, 41.9, 41.5, 36.0, 26.1, 24.0, 23.9, 23.7, 22.9, 19.5; LRMS (EI, 70 eV) *m*/*z* (%) 342 (M⁺, 100); HRMS (EI) calcd for C₂₁H₂₆O₄ (M⁺): 342.1831, found 342.1829.

4.1.9. (3bR,5R,9bS,11S)-5,11-dihydroxy-7-isopropyl-6-methoxy-9bmethyl-3b,4,5,9b,10,11-hexahydrophenanthro[2,1-c]furan-1(3H)-one (14). To a solution of 4 (100 mg, 0.292 mmol) in acetonitrile (4 mL) and water (4 mL), was added ammonium ceric nitrate (320 mg, 0.585 mmol) and the mixture was stirred at room temperature for 3 h. The solvent was evaporated, then CH₂Cl₂ was added and the mixture was washed with water and brine. The organic phase was dried over Na₂SO₄ and concentrated to give a crude product, which was chromatographed on silica gel (20% EtOAc in *n*-hexane) to give pure **14** (84.6 mg, 81%) as a colourless solid, ¹H NMR (CDCl₃, 300 MHz) & 7.27-7.17 (m, 2H), 5.10-5.08 (m, 1H), 5.00-4.81 (m, 2H), 4.76 (d, J=4.5 Hz, 1H), 3.88 (s, 3H), 3.28 (sept, J=5.1 Hz, 1H), 3.12 (br s, 1H), 3.05 (d, J=8.7 Hz, 1H), 2.69 (d, J=11.1 Hz, 1H), 2.18-1.91 (m, 3H), 1.30 (d, J=5.1 Hz, 3H), 1.18 (d, J=5.1 Hz, 3H), 1.16 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 173.6, 165.3, 156.1, 143.6, 139.8, 129.7, 127.1, 126.9, 121.3, 70.7, 62.7, 62.2, 60.3, 41.7, 36.9, 36.2, 27.9, 26.1, 24.3, 23.6, 23.3; LRMS (EI, 70 eV) *m/z* (%) 358 (M⁺, 100); HRMS (EI) calcd for C₂₁H₂₆O₅ (M⁺): 358.1780, found 358.1780.

4.1.10. (3bR,9bS,11S)-7-isopropyl-6-methoxy-11-(methoxymethoxy)-9b-methyl-3b,4,5,9b,10,11-hexahydrophenanthro[2,1-c]furan-1(3H)one (16). To a solution of compound 4 (26 mg, 0.076 mmol) in CH₂Cl₂ (3 mL) were added MOMCl (0.029 mL, 0.38 mol) and N,N-Diisopropylethylamine (0.066 mL, 0.38 mmol). The mixture was stirred at room temperature overnight and water (3 mL) was added. The mixture was washed with water and brine. The organic phase was dried over Na₂SO₄ and concentrated to give a crude product, which was chromatographed on silica gel (7% EtOAc in *n*-hexane) to give pure **16** (26.9 mg, 92%) as a colourless solid, mp 90–92 °C; $[\alpha]_D^{25}$ +35 (c 0.1, CH₂Cl₂); IR (KBr) 3446, 2962, 2935, 2887, 1753, 1675, 1027 cm⁻¹, ¹H NMR (CDCl₃, 400 MHz) δ 7.12 (s, 2H), 5.01 (d, J=7.2 Hz, 1H), 4.84 (m, 3H), 4.59 (m, 1H), 3.73 (s, 3H), 3.54 (s, 3H), 3.29 (sept, J=4.8 Hz, 1H), 3.10–2.89 (m, 2H), 2.78 (d, J=14.4 Hz, 1H), 2.60 (m, 1H), 2.00-1.82 (m, 3H), 1.23 (d, J=4.8 Hz, 3H), 1.22 (d, J=4.8 Hz, 3H), 1.18 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 172.8, 166.0, 155.4, 144.0, 139.2, 127.5, 125.5, 124.2, 120.2, 96.4, 70.1, 65.2, 60.5, 55.8, 41.6, 40.2, 35.7, 26.1, 23.9, 23.9, 23.8, 22.7, 19.5; LRMS (EI, 70 eV) m/z (%) 386 (M⁺, 56), 326 (100); HRMS (EI) calcd for C₂₃H₃₀O₅ (M⁺): 386.2093, found 386.2107.

4.1.11. (3bR,5R,9bS,11S)-5-hydroxy-7-isopropyl-6-methoxy-11-(methoxymethoxy)-9b-methyl-3b,4,5,9b,10,11-hexahydrophenanthro [2,1-c]furan-1(3H)-one (**17**). To a solution of **16** (113 mg, 0.292 mmol) in acetonitrile (4 mL) and water (4 mL), was added ammonium ceric nitrate (320 mg, 0.585 mmol) and the mixture was stirred at room temperature for 5 h. The solvent was evaporated, then CH₂Cl₂ was added and the mixture was washed with water and brine. The organic phase was dried over Na₂SO₄ and concentrated to give a crude product, which was chromatographed on silica gel (20% EtOAc in *n*-hexane) to give pure **17** (104.4 mg, 89%) as a colourless solid, ¹H NMR (CDCl₃, 300 MHz) δ 7.26–7.15 (m, 2H), 5.09 (d, *J*=4.2 Hz, 1H), 5.01 (d, *J*=6.9 Hz, 1H), 4.84 (m, 3H), 4.60 (d, *J*=4.8 Hz, 1H), 3.88 (s, 3H), 3.53 (s, 3H), 3.28 (sept, *J*=7.2 Hz, 1H), 3.10 (br s, 1H), 3.02 (d, *J*=14.4, 5.4 Hz, 1H), 1.30 (d, *J*=7.2 Hz, 3H), 1.18 (d, *J*=7.2 Hz, 3H), 1.16 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 172.7, 166.1, 156.2, 143.7, 139.8, 129.8, 126.9, 125.9, 120.9, 96.5, 70.2, 65.3, 62.7, 62.2, 55.9, 40.3, 37.1, 36.0, 28.0, 26.1, 24.3, 23.7, 23.3; LRMS (EI, 70 eV) *m/z* (%) 402 (M⁺, 44), 309 (100); HRMS (EI) calcd for C₂₃H₃₀O₆ (M⁺): 402.2042, found 402.2028.

4.1.12. (3bR,9bS,11S)-7-isopropyl-6-methoxy-11-(methoxymethoxy)-9b-methyl-3b,4,10,11-tetrahydrophenanthro[2,1-c]furan-1,5 (3H,9bH)-dione (18). To a solution of compound 17 (38 mg, 0.094 mmol) in anhydrous CH₂Cl₂ (3 mL) were added pyridinium dichromate (70.6 mg, 0.188 mol) and molecular sieves 4 A (120 mg). The mixture was stirred at room temperature overnight, diluted with ethyl acetate, and filtered through a pad of silica gel. The filtrate was concentrated under reduced pressure to give a crude product, which was chromatographed on silica gel (15% EtOAc in *n*-hexane) to give pure **18** (36.5 mg, 97%) as a colourless solid, mp 106–108 °C, ¹H NMR (CDCl₃, 400 MHz) δ 7.47 (d, *J*=8.4 Hz, 1H), 7.19 (d, J=8.4 Hz, 1H), 5.01 (d, J=6.8 Hz, 1H), 4.83-4.81 (m, 3H), 4.65 (d, *J*=5.6 Hz, 1H), 3.82 (s, 3H), 3.52 (s, 3H), 3.41 (sept, *J*=7.2 Hz, 1H), 3.08 (dd, *J*=14.0, 7.2 Hz, 1H), 2.80 (m, 3H), 1.97 (dd, *J*=14.4, 5.2 Hz, 1H), 1.29 (s, 3H), 1.25 (d, *J*=7.2 Hz, 3H), 1.19 (d, *J*=7.2 Hz, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 194.5, 172.1, 163.2, 158.4, 150.8, 142.1, 132.0, 126.2, 124.4, 118.7, 96.5, 69.7, 64.4, 62.6, 55.9, 40.1, 39.2, 37.3, 36.1, 25.8, 23.7, 23.1, 23.1; LRMS (EI, 70 eV) m/z (%) 400 (M⁺, 37), 385 (100); HRMS (EI) calcd for $C_{23}H_{28}O_6$ (M⁺): 400.1886, found 400.1871.

4.1.13. (3bR,9bS,11S)-6,11-dihydroxy-7-isopropyl-9b-methyl-3b,4,10,11-tetrahydrophenanthro[2,1-c]furan-1,5(3H,9bH)-dione (19). To a solution of 18 (25 mg, 0.062 mmol) in dichloromethane (2 mL), under nitrogen at -78 °C, was added BBr₃ (0.018 mL, 0.186 mol) and the mixture was stirred at -78 °C for 1 h and warmed to room temperature. An aqueous NaHCO₃ solution (10%) was added and the extracts were washed with brine. The organic phase was dried over Na₂SO₄ and concentrated to give a crude product, which was chromatographed on silica gel (20% EtOAc in nhexane) to give pure **19** (20.7 mg, 97%) as a white solid, ¹H NMR (CDCl₃, 300 MHz) δ 13.04 (s, 1H), 7.44 (d, J=8.1 Hz, 1H), 6.90 (d, J=8.1 Hz, 1H), 4.89–4.78 (m, 3H), 3.36 (sept, *J*=6.6 Hz, 1H), 3.17–3.10 (m, 1H), 2.89–2.73 (m, 3H), 2.62–2.60 (m, 1H), 2.06 (dd, *J*=14.4, 6.3 Hz, 1H), 1.31 (s, 3H), 1.25 (d, *I*=6.6 Hz, 3H), 1.23 (d, *I*=6.6 Hz, 3H); ¹³C NMR (CDCl₃, 100 MHz) § 201.8, 172.9, 162.0, 161.7, 149.1, 135.9, 133.9, 127.5, 114.2, 113.9, 70.1, 59.5, 40.7, 40.6, 36.0, 36.0, 26.1, 23.2, 22.1, 22.0; LRMS (EI, 70 eV) m/z (%) 342 (M⁺, 40), 327 (100); HRMS (EI) calcd for C₂₀H₂₂O₅ (M⁺): 342.1467, found 342.1467.

4.1.14. 7,8- β -Epoxy-2 β ,19-dihydroxy-14-oxo-18 (4 \rightarrow 3)abeo-abieta-3,9,12-trien-18-oic acid lactone (**20**). To a solution of **19** (30 mg, 0.088 mmol) in methanol (2 mL) at 0 °C was added sodium borohydride (3.3 mg, 0.088 mmol) in three portions. After stirring at 0 °C for 30 min, the mixture was quenched with an aqueous NH₄Cl solution (10%) and extracted with EtOAc. The organic layer was washed with brine and dried over Na₂SO₄, and concentrated to give a crude product. The crude product was dissolved in MeOH (3 mL) and a solution of NaIO₄ (19.8 mg, 0.092 mmol) in water (1 mL) was added at 0 °C. After stirring at 0 °C for 50 min, the mixture was extracted with EtOAc. The organic layer was washed with brine and dried over Na₂SO₄, and concentrated to give a crude product, which was chromatographed on silica gel (50% EtOAc in *n*-hexane) to give pure **20** (20.7 mg, 97%) as a yellow solid, ¹H NMR (CDCl₃, 300 MHz) δ 6.97 (d, *J*=6.6 Hz, 1H), 6.44 (d, *J*=6.6 Hz, 1H), 4.79 (s, 2H), 4.70 (d, *J*=6.0 Hz, 1H), 4.12 (d, *J*=5.1 Hz, 1H), 2.92 (sept, *J*=6.9 Hz, 1H), 2.53 (dd, *J*=11.7, 6.0 Hz, 1H), 2.40 (d, *J*=14.4 Hz, 1H), 2.29 (m, 1H), 2.22 (d, *J*=13.5 Hz, 1H), 1.90 (dd, *J*=14.4, 5.7 Hz, 1H), 1.32 (s, 3H), 1.09 (d, *J*=6.9 Hz, 6H); ¹³C NMR (CDCl₃, 100 MHz) δ 194.0, 172.9, 162.5, 149.8, 142.1, 135.0, 126.8, 121.4, 70.1, 66.2, 59.4, 56.8, 43.5, 41.4, 38.0, 26.1, 23.7, 21.6, 21.3, 18.7; LRMS (EI, 70 eV) *m/z* (%) 342 (M⁺, 38), 327 (100); HRMS (EI) calcd for C₂₀H₂₂O₅ (M⁺): 342.1467, found 342.1469.

4.1.15. 2β -Hydroxyl triptonide (**21**). To a solution of compound **20** (218 mg, 0.64 mmol) in acetonitrile (6 mL) was added an aqueous $Na_2(EDTA)$ solution (4×10⁻⁴ M, 6 mL). The resulting homogeneous solution was cooled to 0 °C, followed by addition of 1,1,1-trifluoroacetone (0.3 mL) via a precooled syringe. To this homogeneous solution was added in portions a mixture of sodium bicarbonate (0.13 g, 1.55 mmol) and Oxone (0.308 g, 1.0 mmol) in a period of 1 h (pH 7–7.5). The reaction was monitored by TLC. The mixture was poured into water and extracted with dichloromethane. The extracts were dried (Na₂SO₄), filtered, and concentrated to give a crude product, which was dissolved in MeOH (10 mL) and was added H₂O₂ (30%, 0.5 mL, 5.0 mmol) at room temperature. After stirring for 1 h, the mixture was extracted with EtOAc. The organic layer was washed with brine and dried over Na₂SO₄, and concentrated to give a crude product, which was chromatographed on silica gel (50% EtOAc in *n*-hexane) to give pure **21** (164 mg, 69%) as a white solid, mp 234–236 °C; $[\alpha]_D^{25}$ –30 (*c* 0.05, CHCl₃); IR (KBr) 3270, 2932, 1757, 1731, 1143 cm⁻¹, ¹H NMR (CDCl₃, 300 MHz) δ 4.78 (s, 2H), 4.63 (m, 1H), 4.07 (d, *J*=2.7 Hz, 1H), 3.83 (d, J=2.7 Hz, 1H), 3.42 (d, J=5.1 Hz, 1H), 2.76 (dd, J=12.9, 5.1 Hz, 1H), 2.39 (sept, *J*=6.9 Hz, 1H), 2.26 (m, 1H), 2.12 (d, *J*=13.2 Hz, 1H), 1.89 (d, *J*=13.8 Hz, 1H), 1.54 (dd, *J*=13.5, 5.7 Hz, 1H), 1.26 (s, 3H), 0.97 (d, J=6.9 Hz, 3H), 0.88 (d, J=6.9 Hz, 3H); ¹³C NMR (CDCl₃, 100 MHz) δ 196.7, 172.6, 161.5, 127.0, 70.0, 66.5, 65.1, 60.9, 60.2, 59.2, 58.8, 56.6, 40.9, 38.9, 35.4, 25.8, 22.8, 18.0, 16.2, 15.4; LRMS (EI, 70 eV) m/z (%) 374 (M⁺, 2), 113 (100); HRMS (EI) calcd for C₂₀H₂₂O₇ (M⁺): 374.1366, found 374.1341.

4.1.16. Preparation of tripdiolide (**1**). To a solution of **21** (3.74 mg, 0.01 mmol) and Eu(FOD)₃ (10 mg, 0.01 mmol) in MeOH (0.4 mL) at -78 °C was added NaBH₄ (0.38 mg, 0.01 mmol) in MeOH (0.4 mL). After stirring for 1 h, the mixture was extracted with EtOAc. The organic layer was washed with brine and dried over Na₂SO₄, and concentrated to give a crude product, which was chromatographed on silica gel (50% EtOAc in *n*-hexane) to give pure **1** (1.76 mg, 47%) as a white solid, mp 220–222 °C; $[\alpha]_D^{25}$ –110.2 (*c* 0.170, MeOH); IR (KBr) 3580, 3530, 2911, 1759, 1680, 1449, 1415, 1350, 1230, ¹H NMR (CDCl₃, 400 MHz) δ 4.74 (s, 2H), 4.58 (d, *J*=5.6 Hz, 1H), 3.90 (d, *J*=3.2 Hz, 1H), 3.49 (d, *J*=3.2 Hz, 1H), 3.38 (br s, 1H), 3.34 (d, *J*=5.3 Hz, 1H), 2.74 (br s, 1H), 2.62 (m, 1H), 2.20 (m, 2H), 2.05 (dd, *J*=14.7, 13.1 Hz, 1H), 1.83 (br d, *J*=13.8 Hz, 1H), 1.41 (dd, *J*=13.8,

5.6 Hz, 1H), 1.29 (s, 3H), 0.98 (dd, *J*=6.9 Hz, 3H), 0.85 (dd, *J*=6.9 Hz, 3H); 13 C NMR (CDCl₃, 100 MHz) δ 172.7, 162.0, 126.9, 73.4, 70.1, 66.3, 65.7, 60.8, 60.0, 59.3, 57.4, 54.5, 40.8, 38.1, 35.9, 28.2, 23.3, 17.7, 16.8, 15.3; LRMS (EI, 70 eV) *m*/*z* (%) 376 (M⁺, 0.5); HRMS (EI) calcd for C₂₀H₂₄O₇ (M⁺): 376.1522, found 376.1523.

Acknowledgements

This work was supported by National Science & Technology Major Project 'Key New Drug Creation and Manufacturing Program', China (Number:2009ZX09102-026).

Supplementary data

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.tet.2010.12.014.

References and notes

- Huang, S. The Saint Peasant's Scripture of Materia Medica (18357); Publishing house for Chinese medicine classics: Beijing, republished in 1982; p 309.
- Wang, X. X.; Matta, R.; Shen, G.; Nelin, L. D.; Pei, D. H.; Liu, Y. S. J. Mol. Med. 2006, 84, 405.
- 3. Ding, G. S. Clin. Ther. 1987, 9, 345.
- 4. Kupchan, S. M.; Court, W. A.; Dailey, R. G., Jr.; Gilmore, C. J.; Bryan, R. F. J. Am. Chem. Soc. **1972**, 94, 7194.
- 5. Kupchan, S. M.; Schubert, R. M. Science 1974, 185, 791.
- 6. Lou, Y. J.; Jin, J. Leuk. Lymphoma 2004, 45, 373.
- Panichakul, T.; Wanun, T.; Reutrakul, V.; Sirisinha, S. Asian Pac. J. Allergy Immunol. 2002, 20, 167.
- 8. Tao, X. L.; Cai, J. J.; Lipsky, P. E. J. Pharmacol. Exp. Ther. 1995, 272, 1305.
- Tao, X. L.; Fan, F.; Hoffmann, V.; Gao, C. Y.; Longo, N. S.; Zerfas, P.; Lipsky, P. E. Arthritis Rheum. 2008, 58, 1774.
- Matlin, S. A.; Belenguer, A.; Stacey, V. E.; Qian, S. E.; Xu, Y.; Zhang, J. W.; Sanders, J. K. M.; Amor, S. R.; Pearce, C. M. *Contraception* **1993**, *47*, 387.
- Ma, J.; Dey, M.; Yang, H.; Poulev, A.; Pouleva, R.; Dorn, R.; Lipsky, P. E.; Kennelly, E. J.; Raskin, I. *Phytochemistry* **2007**, 68, 1172.
- 12. Lipsky, P. E.; Tao, X.; Cai, J.; Kovacs, W. J.; Oisen, J. U.S. Patent 5,616,458, 1997. 13. Fidler, J. M.; Li, K.; Chung, C.; Wei, K.; Ross, J. A.; Gao, M.; Rosen, G. D. *Mol.*
- Cancer Ther. 2003, 2, 855. 14. Lin, S.; Deng, S. S.; Que, H. Q.; Guo, S. M.; Qi, Y. P.; Fan, Z. S. Chin. Tradit. Herb.
- Drugs 2008, 39, 1409.
 15. (a) Buckanin, R. S.; Chen, S. J.; Frieze, D. M.; Sher, F. T.; Berchtold, G. A. J. Am. Chem. Soc. 1980, 102, 1200; (b) Van Tamelen, E. E.; Demers, J. P.; Taylor, E. G.; Koller, K. J. Am. Chem. Soc. 1980, 102, 5424; (c) Garver, L. C.; Van Tamelen, E. E. J. Am. Chem. Soc. 1982, 104, 867; (d) Van Tamelen, E. E.; Leiden, T. M. J. Am. Chem. Soc. 1982, 104, 1785; (e) Yang, D.; Ye, X. Y.; Xu, M.; Pang, K. W.; Zou, N.; Letcher, R. M. J. Org. Chem. 1998, 63, 6446; (f) Yang, D.; Ye, X. Y.; Xu, M. J. Org. Chem. 2000, 65, 2208.
- 16. Zhou, B.; Li, X. M.; Feng, H. J.; Li, Y. C. Tetrahedron 2010, 66, 5396.
- 17. Mancuso, A. J.; Swern, D. Synthesis 1981, 1981, 165.
- Tamao, K.; Ishida, N.; Ito, Y.; Kumada, M.; Tran, V. D.; Overman, L. E. Org. Synth. 1990, 69, 96.
- 19. Boulin, B.; Arreguy-San Miguel, B.; Delmond, B. Tetrahedron 1998, 54, 2753.
- (a) Lindgren, B. O.; Nilsson, T. Acta Chem. Scand. **1973**, 27, 888; (b) Kraus, G. A.; Taschner, M. J. J. Org. Chem. **1980**, 45, 1175; (c) Kraus, G. A.; Roth, B. J. Org. Chem. **1980**, 45, 4825; (d) Bal, B. S.; Childers, W. E., Jr.; Pinnick, H. W. Tetrahedron **1981**, 37, 2091.
- 21. Yin, X. Q.; Yang, G. Z.; Li, Y. C. Chin. Chem. Lett. 2001, 12, 579.
- The oxidizing systems examined include PDC, PCC, Collins reagent, Jones reagent, MnO₂, Dess–Martin periodinate, KMnO₄, DDQ, DMSO–(COCl)₂, DMSO–SO₃.Py, Oxone, NaOCl–AcOH, Ca(OCl)₂–TEMPO, NaOCl–TEMPO, and IBX
- 23. Becker, H.; Bremholt, T.; Adler, E. Tetrahedron Lett. 1972, 41, 4205.
- 24. Yang, D.; Wong, M. K.; Yip, Y. C. J. Org. Chem. 1995, 60, 3887.
- 25. Kutney, J. P.; Han, K. Recl. Trav. Chim. Pays-Bas 1996, 115, 77.